General Information	
Academic subject	Food chemistry (I. C. Food Chemistry and Applied Nutrition)
Degree course	Master programme: Food Science and Technology
ECTS credits	3 ECTS
Compulsory attendance	No
Teaching language	Italian

Subject teacher	Name Surname	Mail address	SSD
	Vito Michele	vito.paradiso@uniba.it	CHIM/10
	Paradiso		

ECTS credits details		
Basic teaching activities	2.5 ECTS Lectures	0.5 ECTS Laboratory or field classes

Class schedule	
Period	II semester
Course year	First
Type of class	Lectures, workshops, field classes

Time management		
Hours	75	
In-class study hours	27	
Out-of-class study hours	48	

Academic calendar	
Class begins	March 2 nd , 2020
Class ends	June 12 th , 2020

Syllabus	
Prerequisites/requirements	Principles of general, inorganic and organic chemistry. Principles of food constituents and food technology
Expected learning outcomes	 Knowledge and understanding Knowledge of the main chemical and physical interactions of food constituents Applying knowledge and understanding Skill to apply a systemic approach to the evaluation of food composition and properties Understanding phenomena and constituents determining food quality and its evolution Skill to describe chemical properties and structure of innovative foods Making informed judgements and choices Correctly advising solutions to change properties and quality of food Correctly advising analytical approaches to monitor properties and quality of food Communicating knowledge and understanding Describing chemical constituents and phenomena underlying properties and macroscopic phenomena in food Capacities to continue learning Updating the knowledge of chemical and physical interaction of food constituents
	The expected learning outcomes, in terms of both knowledge and skills, are provided in Annex A of the Academic Regulations of the Degree in Food Science and Technology (expressed through the European Descriptors of the qualification)

Contents	• Water
Contents	Water Food as a dispersed system
	Food as a dispersed systemDispersed systems: emulsions (formation, breaking,
	 Dispersed systems: emulsions (formation, breaking, stabilization), emulsifiers
	Dispersed systems: foams (formation and stability)
	Dispersed systems: gels (structure and properties), gelling
	polymers (pectin, alginates, carragenaans, collagen)
	Dispersed systems: other colloids (gums)
	Lipid oxidation: autoxidation, photoxidation, thermoxidation,
	enzymatic oxidation, antioxidants
	Caramelization and Maillard reaction
	Food pigments
Course program Reference books	a Lacture notes and advectional symplics provided during the
Reference books	 Lecture notes and educational supplies provided during the course.
	• Fennema, O. R. (2010). Dispersed systems. In: Food Chemistry.
	Marcel Dekker.
	Belitz, HD., Grosch, W., & Schieberle, P. (2009). Food
	chemistry. Springer.
	Scientific reviews
Notes	
Teaching methods	Lectures will be presented through PC assisted tools (PowerPoint,
	video). Field and laboratory classes, reading of regulations will be
	experienced.
	Lecture notes and educational supplies will be provided by means of
	online platforms (i.e.: Edmodo, Google Drive)
Evaluation methods	The exam consists of an oral dissertation on the topics developed
	during the theoretical and theoretical-practical lectures in the
	classroom and in the laboratory/production plants, as reported in
	the Academic Regulations for the Master Degree in Food Science and Technology (article 9) and in the study plan (Annex A).
	Students attending at the lectures may have a middle-term
	preliminary exam, consisting of an oral test, relative to the first part
	of the program, which will concur to the final evaluation and will be
	considered valid for a year.
	The evaluation of the preparation of the student occurs on the basis
	of established criteria, as detailed in Annex B of the Academic
	Regulations for the Master Degree in Food Science and Technology.
	Non-Italian students may be examined in English language,
	according to the aforesaid procedures.
Evaluation criteria	Knowledge and understanding
	Describing the main chemical and physical interactions of
	food constituents
	Applying knowledge and understanding
	Describing phenomena and constituents determining the quality of foods dealt with during lessons, as well as the
	quality of foods dealt with during lessons, as well as the evolution of quality during time
	Making informed judgements and choices
	 Expressing reasonable hypotheses about solutions to
	change properties and quality of foods dealt with during
	lessons
	Communicating knowledge and understanding
	 Describing chemical constituents and phenomena
	underlying properties and macroscopic phenomena in
	anderlying properties and macroscopic phenomena in
	foods dealt with during lessons

	 Expressing reasonable hypotheses about the evaluation of chemical properties of foods dealt with during lessons
Receiving times	Tutorial activities: from Monday to Thursday 9.00 a.m. – 12.30 p.m.
	by appointment only